H Kaur Book Spectroscopy Pdf Download EXCLUSIVE

unfortunately, raman spectroscopy has not been used for quantification of medicines, although efforts are underway to overcome this challenge. some of the first efforts to use raman spectroscopy to quantify medicines have focused on pharmaceutical formulations, and the majority of these efforts have focused on determining the amount of active ingredients in tablets (kaur et al., 2010 ; lu et al., 2012 ; nyadong et al., 2009 ; witkowski, 2005). lasers in dentistry. dentistry, vol. 33, no. 1, pp. 1-8, feb. 1993. quench-controlled uv-vis spectroscopy. spectrochimica acta, vol. a, no. 41, pp. 1231-1238, nov. 1986. vibrational circular dichroism and electronic circular dichroism of chiral compounds. b, no. 53, pp. 99-105, feb. 1998. molecular spectroscopy of chiral compounds. 67-75, feb. near-infrared spectroscopy of chiral compounds. 73-78, feb. raman spectroscopy may be used to differentiate the species, similar in composition but differing in structural arrangement, using the rotational-vibrational mode of the constituent molecules. the assignment of the raman bands to specific molecules may thus be used to characterize the molecules of interest in terms of their structures. the presence of excipient materials in a pharmaceutical dosage forms can be assessed using raman spectroscopy. when the excipients are concentrated in the form of a solid matrix, as in tablets and capsules, they can be detected by scanning the surface of the solid materials with the laser. titier p, et al. molecular spectroscopy in pharmaceutical analysis, in analytical chemistry and validated analytical methods, ed. franco simo, (amsterdam, london, new york, san francisco: elsevier science, (1994) 775-782.

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kaur and colleagues developed a novel hplc technique that uses photodiode array detection to screen for and quantify the presence of each excipient in a drug (kaur et al., 2006 ; kaur et al., 2010). when combined with mass spectroscopy, this technique can identify and identify specific excipients, which can provide valuable information on a potential drug's origin. near-infrared spectroscopy can be coupled with thermal-ionization detectors or mass spectrometry to analyze the composition of a drug. it can detect and identify the presence of many components, including the active ingredients, excipients, or contaminants, and quantify their concentrations in the drug (plich et al., 2008). uv-visible spectroscopy can be used to measure the specific wavelengths of light absorbed by an unknown mixture, but it is also useful for identifying the presence and composition of impurities and reacting with other compounds (mclendon and messiaen, 2005). uv spectroscopy has been used to detect forgeries of the ancient past (mckee and shumway, 2000), document authenticity and potential forgeries (mckee and shumway, 2000), and identify the species in dna (palladino et al., 2004). nmr spectroscopy is also useful for identifying chemical species by the way they absorb and/or emit their own unique magnetic field (kaur et al., 2010). the "signature" of individual compounds in a mixture may be either broad or sharp. a broad "signature" of a whole mixture is an indication of the presence of many chemically different components, while a sharp "signature" indicates the chemical similarity of the components. nmr spectroscopy is also useful for analyzing the purity and identity of pharmaceuticals. nmr can identify both active and inactive chemical species in a mixture (girgis et al., 2002). 5ec8ef588b

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